

SYNTHESIS OF 6H-PYRIDO[4,3-b]CARBAZOLES

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Abstract: A few 6H-pyrido[4,3-b]carbazoles were prepared for cytotoxicity testing in cultures of human lung cancer cells. Methyl 6-methoxyindoleacetate **13**, prepared by reduction of methyl 4-(2-nitro-5-methoxyphenyl)-3-oxobutyrate **11**, was condensed with 3-acetylpyridine to give the vinyl indole **14** which was quarternized with p-nitrobenzyl bromide. Cyclization of this salt with the aid of sodium methoxide and ethyl nicotinate methiodide gave the 6H-pyrido[4,3-b]carbazole **16**. Brief treatment with tributyl phosphine in boiling DMF gave the ester **17** which, after reduction with LiAlH₄ followed by treatment with methyl isocyanate, gave 5-hydroxymethyl-9-methoxy-11-methyl-6H-pyrido[4,3-b]carbazole N-methylcarbamate **4**.

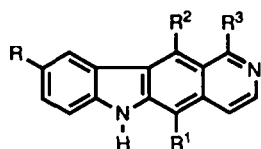
9-Methoxyolivacine **19** was oxidized to the aldehyde **22** with SeO₂. Reduction to the carbinol **24** proceeded smoothly with NaCNBH₃ in acid medium. Treatment with methyl isocyanate gave the carbamate **6** and the acetate **7** was prepared by acetylation of the carbinol **24**.

In a previous paper¹ we reported the synthesis and antitumor activity of 5-hydroxymethyl-11-methyl-6H-pyrido[4,3-b]carbazole N-methylcarbamate, **1**, and suggested a molecular mechanism to account for the antitumor activity of ellipticine **2** which differed from the one proposed by Auclair and Paoletti.² Since 9-methoxyellipticine, **3**, is active in a variety of experimental tumor systems,³ we decided to prepare and test the 9-methoxy derivative **4**.⁴

In our suggested mechanism we postulated that the methyl group at C-5 in ellipticine is the site of metabolic activation but there is no compelling evidence to rule out activation at the alternate C-11 methyl group of **2**. Since a successful synthesis of **5** has proven to be elusive thus far we decided to prepare the olivacine analogue, **6**.⁵ In this paper we report the synthesis of the 6H-pyrido[4,3-b]carbazoles, **4**, **6** and **7**.

The synthesis of **4** was carried out using the same method described for the preparation of **1** with a few significant modifications (Scheme 1).

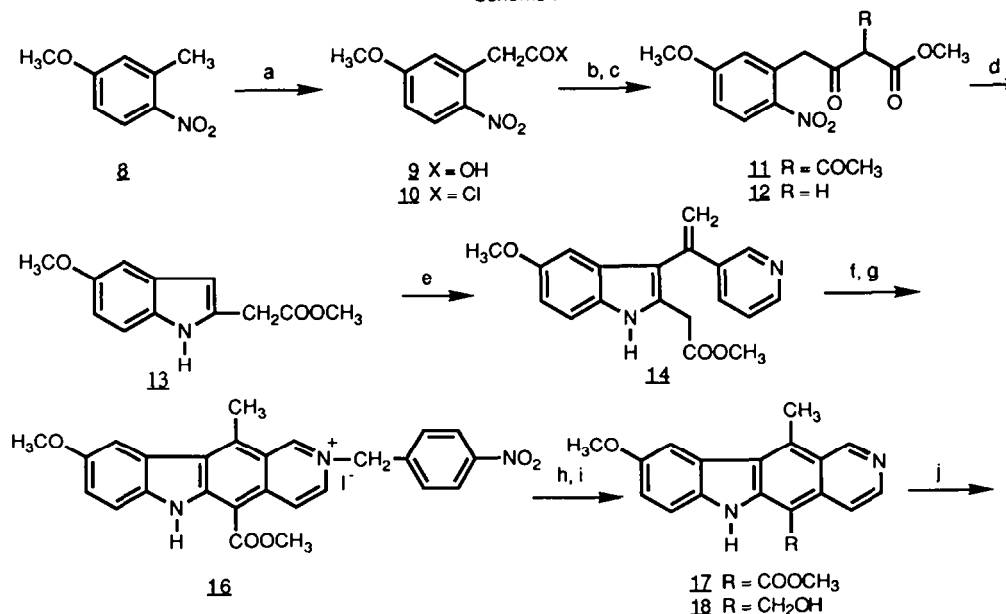
Commercially available 3-methyl-4-nitroanisole **8** was treated with ethyl oxalate and NaH to give the pyruvate ester which, on oxidation with H₂O₂ in NaOH solution, gave after acidification, the acid **9**.⁶ Treatment with SOCl₂ gave **10** which was condensed with the anion prepared from methyl acetoacetate and NaH gave **11**. Ammonolysis furnished the ester **12** which was then reduced with Pd/C and ammonium formate to give **13**. Condensation of the ester **13** with 3-acetylpyridine in the presence of H₂SO₄ gave the vinylindole **14**, quaternization of which with p-nitrobenzyl bromide gave the quaternary salt **15**. Cyclization and aromatization of **15** to **16** was carried out in the presence of NaOCH₃ followed by treatment with ethyl nicotinate methiodide. The methiodide salt was found to give more consistent yields of **17** than treatment with ethyl nicotinate methobromide.



- 1 R = H, R¹ = CH₂OOCNHCH₃, R² = CH₃, R³ = H
- 2 R = R³ = H, R¹ = R² = CH₃
- 3 R = OCH₃, R¹ = R² = CH₃, R³ = H
- 4 R = OCH₃, R¹ = CH₂OOCNHCH₃, R² = CH₃, R³ = H
- 5 R = R³ = H, R¹ = CH₃, R² = CH₂OOCNHCH₃
- 6 R = OCH₃, R¹ = CH₃, R² = H, R³ = CH₂OOCNHCH₃
- 7 R = OCH₃, R¹ = CH₃, R² = H, R³ = CH₂OOCCH₃

Dequaternization using nitrosodimethylaniline and sodium methoxide, which was successful in the synthesis of **1**, gave **17** in poor yield, owing to the difficulty in separating the desired ester from the nitrone formed in the reaction. Heating **16** with triphenylphosphine in DMF⁷ did not give a clean reaction product owing in part to the formation of triphenylphosphine oxide which was difficult to remove. It is known that triphenylphosphine reduces nitro groups to amines and forms triphenylphosphine oxide.⁸ The most successful procedure consisted of heating **16** in boiling DMF with four equivalents of tributylphosphine for a short period of time. Excess reagent and solvent were easily removed by distillation and the product tributylphosphine oxide was separated by chromatography. Reduction of **17** furnished the alcohol **18**.

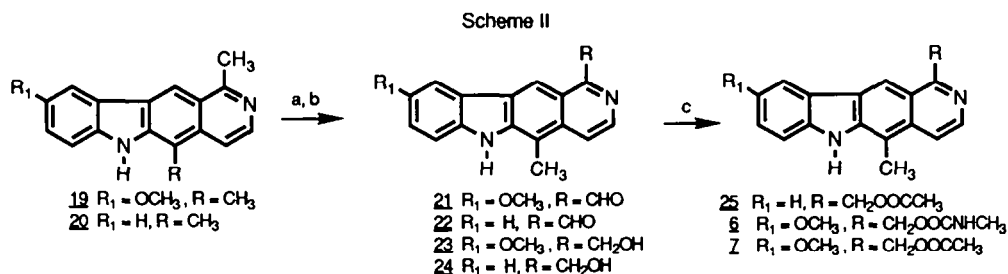
Scheme I



a: (i) (COOCH₃)₂, NaH (ii) H₂O₂, NaOH; b: (i) SOCl₂ (ii) CH₃COCH₂COOCH₃, NaH; c: NH₄OH, MeOH; d: Pd/C, decalin; e: 3-acetylpyridine, H₂SO₄; f: p-nitrobenzyl bromide; g: NaOCH₃, MeOH, ethyl nicotinate methiodide; h: n-Bu₃P, DMF; i: LiAlH₄; j: CH₃NCO, DMAP.

When the carbinol **18** was stirred with CH_3NCO and 4-dimethylaminopyridine (DMAP) in CH_2Cl_2 , the cloudy suspension clarified while being stirred overnight. The carbamate **4** was obtained, but the NMR spectrum indicated that the material was a mixture of isomers formed in the ratio of 2:1. When a DMSO solution of this mixture was gradually heated to 65°C , one of the isomers was converted to the other and upon cooling reversion to the original mixture did not occur. Attempts to separate the isomers has been unsuccessful thus far. When the reaction was carried out in CHCl_3 the isomer ratio was 20:1.⁹

Because attempts to prepare olivacine **20** by a modification of Moody's short synthesis of ellipticine¹⁰ failed,¹¹ we resorted to the synthesis of 9-methoxyolivacine described by Besselievre and Husson.¹²⁻¹⁴ The conversion of **19** and **20** to the carbamate **6** and the acetate **7** and **25** is shown in scheme II.



a: SeO_2 , b: NaCNBH_3 ; c: CH_3NCO or acetic anhydride.

Oxidation of **19** and **20** with SeO_2 gave the aldehydes **21** and **22** as deep red crystalline solids.¹⁵ Reduction with either NaBH_4 or LiAlH_4 was unsuccessful but NaCNBH_3 in acid solution gave the desired carbinols **23** and **24**. Treatment of **24** with CH_3NCO in CH_2Cl_2 in the presence of DMAP gave the carbamate **6** as a single isomer. The acetates **25** and **7** were obtained by acetylation of the respective carbinols **23** and **24**.

Compounds **4**, **6**, **7** and **25** were tested for cytotoxicity against four human lung cancer cell lines in tissue culture at the Albany Medical College by Dr. J. R. Ruckdeschel and Mr. R. Portuese. The results will be reported elsewhere. Suffice it to say that the carbamates **4** and **6** appear to be more cytotoxic than either ellipticine or adriamycin which were used as standards.

EXPERIMENTAL

All reactions were carried out in an atmosphere of dry nitrogen and in flame-dried glassware. Tetrahydrofuran and dioxane were distilled over sodium-benzophenone before use. CH_3NCO and CF_3COOH were distilled before use and all other starting materials were used as received from the supplier. Infrared spectra were recorded on a Perkin-Elmer Model 298 spectrometer. Proton NMR spectra were run on a Varian XL-200 (200 MHz) spectrometer using $(\text{CH}_3)_4\text{Si}$ as the internal standard and are reported in parts per million. Mass spectra were obtained on a Hewlett-Packard Model 5987A GC/MS spectrometer using isobutane as the CI gas. Melting points were taken on a Mel-temp apparatus and are uncorrected. Elemental analyses were performed by Atlantic Microlab Inc., Norcross, GA. Despite careful drying some 6H-pyrido[4,3-b]carbazoles retained H_2O tenaciously. Proton signals attributed to H_2O were observed in the NMR spectra.

5-Methoxy-2-nitrophenylacetic acid, 9: Ten grams (0.06M) of 3-methyl-4-nitroanisole **8** was added to a suspension obtained by mixing 9.1 mL (0.067M) of diethyl oxalate and 3.9g (0.07M) of CH_3ONa in 50 mL of dry ether. The mixture was stirred under reflux for 10 h and the thick orange suspension was treated with H_2O . The resulting solution was treated alternately with 30-32% H_2O_2 and 10 N NaOH. The resulting suspension was filtered to remove unreacted 3-methyl-4-nitroanisole (20%) and the filtrate was cooled to 5°C before being acidified with conc. HCl. The white solid that separated was collected on a filter, washed thoroughly with H_2O and dried. Wt. 9.2g (72%), m.p. 174-176°C (lit value:⁶ 174-176°C (KBr) λ 3140-2740, 2660, 1715, 1620, 1580, 1510, 1410, 1340, 1320, 1290, 1260, 1200, 1175, 1090, 1040, 955, 830, 760, 710, 630 cm^{-1} ; NMR (DMSO-d_6) δ 8.13 (1H, d, J=8.8, H₃), 7.10-7.04 (2H, m, H₄, H₆), 3.97 (2H, s, CH₂), 3.86 (s, OCH₃); MS m/e 212 (M+1).

Methyl (5-Methoxy-2-nitrobenzoyl)-acetoacetate, 11: A suspension of 10.0g (0.047M) of the acid **9**, 3.6 mL of SOCl_2 , 0.5 mL of dry DMF in 125 mL of dry toluene was stirred at room temperature overnight and then warmed at 50°C to give a clear solution. A solution of 10.7 mL (0.1 M) of methyl acetoacetate in 50 mL of THF was added dropwise over a period of 40 min. to a stirred cold (0-5°C) suspension of 2.4g (0.1 M) of NaH in THF. The resulting solution was added to a cooled solution of the acid chloride, **10**, and the resulting mixture was allowed to stir overnight. Water was added and the organic layer was separated. The aqueous layer was extracted with ether (3 x 100 mL) and the combined organic layers were washed with H_2O and dried over Na_2SO_4 . The filtered organic layer was concentrated to dryness and the residue was dissolved in 25 mL of $\text{Et}_2\text{O-CH}_3\text{OH}$ (1:1). On cooling at -20°C overnight the crystals that separated were collected and dried. Wt. 12.6g (82%), suitable for use in the next step. The analytical sample melted at 78-79°C after crystallization from CH₂Cl₂. IR (KBr) λ 2960, 2850, 1710, 1620-1390, 1350-1250, 1190, 1090, 1035, 950, 910, 890, 840, 760, 730, 630 cm^{-1} ; NMR (CDCl_3) δ 8.19 (1H, d, J=9.0, H₃); 6.90 (1H, dd, J=2.8, 9.0, H₄), 6.77 (1H, d, J=2.6, H₆), 4.50 (2H, s, CH₂), 3.99 (3H, s, OCH₃), 3.84 (3H, s, COOCH₃), 2.40 (3H, s, COCH₃); MS m/e 310 (M+1).

Anal. Calcd. for $\text{C}_{14}\text{H}_{15}\text{NO}_7$: C, 54.37; H, 4.85; N, 4.53. Found: C, 54.45; H, 4.90; N, 4.58.

Methyl-4-(5-Methoxy-2-nitrophenyl)-3-oxobutylate, 12: Ten milliliters of NH_4OH was added to a suspension of 6.5g (0.41 M) of the keto-ester **11** in 100 mL of CH_3OH . The resulting mixture was stirred at room temperature until the reaction was complete (30-45 min) as monitored by TLC. The white solid was collected, wt. 6.5g. Concentration of the filtrate gave an additional 1.65g of **12** for a combined yield of 8.15g (75%) suitable for use in the next step. After crystallization from CH_2Cl_2 the analytical sample melted at 85-87°C. IR (KBr) λ 3470, 3130, 2980, 2850, 710, 1590, 1500, 1440, 1400, 1200, 1010, 865, 850, 760, 750, 700, 630, 620 cm^{-1} ; NMR (CDCl_3) δ 8.19 (1H, d, J=9.2, H₃), 6.89 (1H, dd, J=2.8, 9.2, H₄), 6.77 (1H, d, J=2.8, H₆), 4.20 (2H, s, ArCH₂), 3.87 (3H, s, OCH₃), 3.77 (3H, s, COOCH₃), 3.66 (2H, s, CH₂); MS m/e 268 (M+1).

Anal. Calcd. for $\text{C}_{12}\text{H}_{13}\text{NO}_6$: C, 53.93; H, 4.87; N, 5.24. Found: C, 53.97; H, 4.90; N, 5.22.

Methyl-5-Methoxyindole-2-acetate, 13: To a suspension of 6.0g (0.024 M) of **13** in 180 mL of dry CH_3OH there was added 650 mg of 10% Pd/C followed by 17.5g (0.278 M) of ammonium formate. After stirring at room temperature for 15 min. the reaction mixture was filtered (Celite) and the filtrate was evaporated to dryness to leave a residue which was triturated with H_2O , filtered and dried. Wt. 4.82g (90%). After crystallization from CH_3OH the analytical sample melted at 96-98°C. IR (KBr) λ 3370, 3020, 2970, 2840, 1720, 1595, 1485, 1445, 1395, 1330, 1205, 1120, 1035, 1010, 980, 950, 845, 770, 735, 695, 625 cm^{-1} ; NMR (CDCl_3) δ 8.53 (1H, s, NH), 7.23 (1H, d, J=4.0, H₅), 7.01 (1H, d, J=2.4, H₄), 6.81 (1H, dd, J=2.4, 8.8, H₆), 6.27 (1H, d, J=1.0, H₃), 3.82 (3H, s, OCH₃), 3.82 (2H, s, CH₂), 3.74 (3H, s, COOCH₃); MS m/e 220 (M+1).

Anal. Calcd. for $\text{C}_{12}\text{H}_{13}\text{NO}_3$: C, 65.74; H, 5.98; N, 6.39. Found: C, 65.81; H, 6.02; N, 6.39.

1-[2-(Carbomethoxymethyl)-3-(5-methoxyindole)]-1-(3-pyridyl)-ethene, 14: To a solution of 11.41g (0.051 M) of the keto-ester **13** and 18.7 mL (0.17 M) of 3-acetylpyridine in 310 mL of CH_3OH , there was added dropwise with stirring 26.7 g

conc. H₂SO₄. The mixture was refluxed for 6.5 h, cooled and poured into ice-water. The solution was brought to pH 8.5-9.0 by addition of NH₄OH and the solid that separated was collected, washed with H₂O and dried. After crystallization from CH₃OH there was obtained 11.03g (67%) of the desired vinyl indole **14**, m.p. 163-164°C. IR (KBr) λ 3210-2810, 1730, 1610, 1590, 1470-1440, 1295, 1250, 1205, 1170, 1115, 1065, 1030, 1015, 915, 845, 815, 730, 715, 675, 645 cm⁻¹; NMR (CDCl₃) δ 8.82 (1H, s, NH), 8.70 (1H, d, J=2.0), 8.53 (1H, d, J=4.6), 7.62-7.18 (1H, m), 6.82 (1H, dd, J=2.3, 8.8), 6.57 (1H, s), 5.82 (1H, s, =CH), 5.42 (1H, s, =CH), 3.72 (2H, s, CH₂), 3.71 (3H, s, OCH₃), 3.66 (3H, s, COOCH₃); MS m/e 323 (M+1).

Anal. Calcd. for C₁₉H₁₈N₂O₃: C, 70.79; H, 5.63; N, 8.69. Found: C, 70.67; H, 5.69; N, 8.65.

The corresponding p-nitrobenzyl pyridinium bromide **15** was prepared by allowing 3.95g (0.012 M) of the above vinyl indole to react with 8.1 g (0.037 M) of 4-nitrobenzyl bromide in 100 mL of dry acetone overnight and then filtering the quaternary salt **15**. After washing with fresh acetone and drying the salt melted at 190-193°C (dec.). Wt. 6.05g (92%). IR (KBr) λ 3340, 3010, 2960, 1740, 1630, 1575, 1485, 1350, 1295, 1210, 1160, 1070, 945, 810, 710 cm⁻¹; NMR (DMSO-d₆) δ 11.43 (1H, s, NH), 9.23 (1H, d, J=5.8), 9.14 (1H, s), 8.56 (1H, d, J=8.0), 8.23-8.17 (2H, m), 7.72 (1H, d, J=8.6), 7.33 (1H, d, J=9.0), 6.73 (1H, dd, J=2.0, 8.8, H₆), 6.15 (1H, s, =CH), 6.03 (1H, s, =CH), 5.69 (2H, s, CH₂N⁺), 3.83 (2H, s, CH₂COOCH₃), 3.63 (3H, s, OCH₃), 3.44 (3H, s, COOCH₃).

Anal. Calcd. for C₂₆H₂₄BrN₃O₅: C, 58.00; H, 4.49; N, 7.81. Found: C, 58.09; H, 4.53; N, 7.78.

2-(4-Nitrobenzyl)-5-carbomethoxy-9-methoxy-11-methyl-6H-pyrido[4,3-b]carbazolium iodide, 16: To a solution of 8.00g (0.015 M) of the salt **15** in 800 mL of dry CH₃OH there was added 16.0g (0.055 M) of ethyl nicotinate methiodide. A solution of CH₃ONa prepared by dissolving 860 mg (0.037 M) of Na in 90 mL of CH₃OH was added and the resulting dark red solution was allowed to stir at room temperature for 3 h during which time an orange solid separated. The mixture was left overnight, filtered and the solid was washed with CH₃OH to leave 8.2g (95%) of **16**, m.p. 268-270°C (dec.). It was used directly in the next step. IR (KBr) λ 3440, 2960, 1680, 1625, 1600, 1480, 1345, 1200, 1070, 1030, 865, 810, 740 cm⁻¹.

5-Carbomethoxy-9-methoxy-11-methyl-6H-pyrido[4,3-b]carbazole, 17: A mixture of 3.5g (6mM) of **16**, and 5.05 mL (0.184 M) of tributylphosphine in 50 mL of dry DMF was refluxed for 15 min and cooled to room temperature. The volatiles were removed by distillation *in vacuo* and the residue was washed with hexane (2 x 50 mL). The residual dark oil was chromatographed on a silica gel column using CH₂Cl₂: CH₃OH (98:2) as the eluant. The solid that was obtained was crystallized from ethyl acetate to give 1.01g (52%) of **17**, m.p. 178-180°C; IR (KBr) λ 3400, 2960, 1675, 1605, 1480, 1445, 1325, 1300, 1215, 1070, 1040, 875, 840, 790, 700 cm⁻¹; NMR (CDCl₃) δ 10.34 (1H, s, NH), 9.60 (1H, brs, H₁), 8.81 (1H, d, J=6.3, H₃), 8.58 (1H, brs, H₄), 7.71 (1H, d, J=2.3, H₁₀), 7.35 (1H, d, J=8.6, H₇), 7.14 (1H, dd, J=2.4, 8.7, H₈), 4.11 (3H, s, OCH₃), 3.95 (3H, s, COOCH₃), 3.16 (3H, s, C-11, CH₃); MS m/e 321 (M+1).

Anal. Calcd. for C₁₉H₁₆N₂O₃: C, 71.23; H, 5.04; N, 8.75. Found: C, 71.11; H, 5.12; N, 8.68.

5-Hydroxymethyl-9-methoxy-11-methyl-6H-pyrido[4,3-b]carbazole, 18: The ester **17** was added portionwise to a stirred suspension of 0.86g (0.023 M) of LiAlH₄ in 350 mL of THF and the reaction mixture was stirred at room temperature for 4 h before being quenched by the addition of 1.5 mL of H₂O. Then 200 mL of CHCl₃ was added and after stirring the suspension was filtered. The filtrate was evaporated to dryness to leave 1.22g of the carbinol **18**. The solid that remained on the filter was washed thoroughly with warm CHCl₃. Evaporation of the dried CHCl₃ extracts gave an additional 0.46g of **18** making the total 1.68g (74%). Recrystallization from CH₃OH afforded the analytical sample, m.p. 209-211°C (dec.). IR (KBr) λ 3300-2840, 1605, 1485, 1410, 1300, 1260, 1220, 1150, 1040, 990, 820 cm⁻¹; NMR (DMSO-d₆) δ 11.28 (1H, s, NH), 9.69 (1H, s, H₁), 8.41 (1H, d, J=6.0, H₃), 8.05 (1H, d, J=6.0, H₄), 7.88 (1H, d, J=2.2, H₁₀), 7.51 (1H, d, J=8.7, H₇), 7.19 (1H, dd, J=2.4, 8.7, H₈), 5.22 (2H, s, CH₂OH), 3.91 (3H, s, OCH₃), 3.28 (3H, s, C-11-CH₃); MS m/e 293 (M+1).

Anal. Calcd. for $C_{18}H_{16}N_2O_2$: C, 73.95; H, 5.52; N, 9.59. Found: C, 73.83; H, 5.57; N, 9.56.

5-Hydroxymethyl-9-methoxy-11-methyl-6H-pyrido[4,3-b]carbazole N-methylcarbamate, 4: (a) To a suspension of 100 mg (0.342 mM) of **18** in 100 mL of dry $CHCl_3$ containing 50 mg of dissolved DMAP there was added 0.5 mL of CH_3OH and the suspension was stirred overnight. The resulting clear solution was washed with a phosphate buffer, pH 6 to remove the DMAP. After washing with H_2O and drying over Na_2SO_4 the solution was evaporated to dryness. An NMR spectrum of the crude carbamate revealed very little of the isomer with a CH_2OH signal at 5.59 ppm. Chromatography on a silica column using $CHCl_3/CH_3OH$ (10:1) as the eluant gave 62 mg (53%) of the desired carbamate, **4**, m.p. 210-215°C (dec.) (KBr) λ 3340-3300, 2960, 1695, 1610, 1530, 1480, 1220, 1155-1145, 1050, 995, 810 cm^{-1} : NMR (DMSO- d_6) δ 11.45 (s, NH), 9.72 (1H, s, H_1), 8.44 (1H, d, $J=6.1$, H_3), 7.94 (1H, d, $J=6.0$, H_4), 7.88 (1H, d, $J=2.3$, H_{10}), 7.50 (1H, d, $J=8.6$, H_7), 7.21 (1H, dd, $J=2.4$, 8.7, H_8), 7.04 (1H, m, NH- CH_3), 5.74 (2H, s, CH_3), 3.91 (3H, s, OCH $_3$), 3.30 (3H, s, C-11, CH_3), 3.00 (3H, d, $J=4.7$, NH CH_3).

Anal. Calcd. for $C_{20}H_{19}N_3O_3 \cdot H_2O$: C, 65.38; H, 5.76; N, 11.42. Found: C, 65.62; H, 5.48; N, 11.34.

(b). A similar run was carried out in 50 mL of dry CH_2Cl_2 using 50 mg of **18**, 20 mg of DMAP and 0.25 mL of CH_3NCO . TLC of the resulting carbamate showed two spots with similar R_f values. All attempts to separate this mixture were unsuccessful. NMR spectrum showed two sets of peaks with the methylene proton signals being at δ 5.74 and 5.59 in a ratio of 2:1. A variable temperature 1H NMR study showed that as the temperature increased the upfield signal at 5.59 ppm gradually diminished until at 65°C the entire NMR spectrum showed that only one isomer was present. On cooling to 25°C the missing signals did not re-appear.¹⁰

Anal. Calcd. for $C_9H_{15}NO_2 \cdot 0.1 H_2O$: C, 63.15; H, 8.83; N, 8.19. Found: C, 63.02; H, 8.78; N, 8.22.

1-Formyl-5-methyl-9-methoxy-6H-pyrido[4,3-b]carbazole, 21: A suspension of 1.4 g (5.0 mM) of methoxyolivacine **19** in 300 mL of dry dioxane was heated under reflux until a clear solution resulted. Then 620 mg (5.0 mM) of SeO_2 was added and refluxing was continued for 5 h. The hot reaction mixture was filtered through a bed of Celvolite. The solid was washed with $CHCl_3$ and the combined filtrates were evaporated to dryness. The residue was chromatographed on silica gel using $CH_2Cl_2-CH_3OH$ (19:1) as the eluant. There was obtained 1.02 g (69%) of a deep red crystalline solid, m.p. 250-258°C (lit. value:¹⁵ m.p. 270°C). IR (KBr) λ 3300, 2840, 1690, 1625, 1585, 1490, 1410, 1310, 1180, 1040, 880, 845, 815, 770, 670, 620 cm^{-1} : NMR (DMSO- d_6) δ 11.37 (1H, s, NH), 10.35 (1H, s, H_{11}), 9.83 (1H, s, CHO), 8.69 (1H, d, $J=6.0$, H_3), 8.29 (1H, d, $J=5.8$, H_4), 7.85 (1H, d, $J=2.0$, H_{10}), 7.48 (1H, d, $J=9.0$, H_7), 7.19 (1H, dd, $J=8.8$, H_8), 3.91 (3H, s, OCH $_3$), 3.34 (s, H_2O), 2.86 (3H, s, CH_3); MS m/e 291 (M+1).

1-Formyl-5-methyl-6H-pyrido[4,3-b]carbazole, 22: Oxidation of olivacine **20** was carried out in the same way as described for the oxidation of **19**, in 46% yield, m.p. 288-290°C (dec.). IR (KBr) λ 3260-2630, 1695, 1615, 1590, 1510, 1255, 1215, 1155, 880, 745 cm^{-1} : NMR (DMSO- d_6) δ 11.57 (1H, s, NH), 10.36 (1H, s, H_{11}), 9.86 (1H, s, CHO), 8.72 (1H, d, $J=6.0$, H_3), 8.33-8.30 (2H, m), 7.57 (1H, d, $J=3.6$), 7.31-7.25 (2H, m), 3.32 (s, H_2O), 2.89 (3H, s, CH_3); MS m/e 261 (M+1).

Anal. Calcd. for $C_{17}H_{12}N_2O \cdot 0.6 H_2O$: C, 75.27; H, 4.87; N, 10.33. Found: C, 75.20; H, 4.83; N, 10.95.

1-Hydroxymethyl-9-methoxy-5-methyl-6H-pyrido[4,3-b]carbazole, 23: A suspension of 0.5 g (1.72 mM) of aldehyde **21** in 100 mL of abs. C_2H_5OH containing 4 mL of CF_3COOH was stirred at room temperature until a clear solution resulted. Then 1.6 g (25.5 mM) of $NaCNBH_3$ was added and stirring was continued for 30 min. The mixture was concentrated to 50 mL and then 200 mL of $CHCl_3$ was added. The organic layer was washed with $NaHCO_3$ solution followed by H_2O and then brine. Concentration of the dried $CHCl_3$ left 0.48 g (95%) of almost pure carbinol, **24**, suitable for use in the next step. An analytical sample was obtained after chromatography on silica gel using $CHCl_3-CH_3OH$ as the developing solvent, m.p. 198-201°C. IR (KBr) λ 3400-3080, 2920, 1620, 1480, 1395, 1285, 1200, 1030, 810, 765 cm^{-1} .

NMR (DMSO- d_6) δ 11.18 (1H, s, NH), 8.97 (1H, s, H₁₁), 8.32 (1H, d, J=6.0, H₃), 7.92-7.86 (2H, m, H₄, H₂), 7.44 (1H, d, J=8.6, H₇), 7.13 (1H, dd, J=2.6, 8.6, H₈), 5.34 (1H, t, OH), 5.20 (2H, d, J=5.2, CH₂OH), 3.89 (3H, s, OCH₃), 3.31 (s, H₂O), 2.80 (3H, s, CH₃); MS *m/e* 293 (M+1).

Anal. Calcd. for C₁₈H₁₆N₂O₂·0.25 H₂O: C, 72.84; H, 5.48; N, 9.44. Found: C, 72.81; H, 5.59; N, 9.09.

1-Hydroxymethyl-5-methyl-6H-pyrido[4,3-b]carbazole, 24: Reduction of **22** was carried out as described above in 75% yield, m.p. 240-243°C (dec.). IR (KBr) λ 3340-3060, 1605, 1388, 1255, 1230, 1060, 870, 815, 750, 725 cm⁻¹; NMR (DMSO- d_6) δ 8.97 (1H, s, NH), 8.37-8.30 (3H, m), 7.91 (1H, d, J=6.4), 7.55-7.25 (3H, m), 5.22 (2H, s, CH₂OH), 3.18 (s, H₂O), 2.89 (3H, s, CH₃); MS *m/e* 263 (M+1).

Anal. Calcd. for C₁₇H₁₄N₂O·0.5 H₂O: C, 75.28; H, 5.53; N, 10.37. Found: C, 75.19; H, 5.62; N, 10.29.

1-(Hydroxymethyl-9-methoxy-5-methyl-6H-pyrido[4,3-b]carbazole N-methylcarbamate, 6: A suspension of 340 mg (1.16 mM) of **24**, 260 mg of DMAP in 180 mL of CH₂Cl₂ was stirred at room temperature for 30 min. before 2.1 mL of CH₃NCO was added. The mixture was stirred for 18 h and then concentrated to about 10 mL. The clear solution was cooled in an ice-salt bath for 2 h and the yellow crystals which separated were washed with cold (-20°C) CH₂Cl₂. After drying there was obtained 215 mg (54%) of **6**, m.p. 225-228°C (dec.). IR (KBr) λ 3320, 3220, 1690, 1605, 1490, 1415, 1280, 1210, 1150, 1040, 870, 800, 765 cm⁻¹; NMR (DMSO- d_6) δ 11.21 (1H, s, NH), 8.88 (1H, s, H₁₁), 8.33 (1H, d, J=6.0, H₃), 7.95-7.89 (2H, m, H₄, H₁₀), 7.45 (1H, d, J=8.8, H₇), 7.18-7.12 (2H, m, H₈, NHCH₃), 5.74 (2H, s, CH₂), 3.89 (3H, s, OCH₃), 3.32 (s, H₂O), 2.81 (3H, s, C-5 CH₃), 2.61 (3H, d, J=4.4, NHCH₃); MS *m/e* 350 (M+1).

Anal. Calcd. for C₂₀H₁₉N₃O₂·0.67 H₂O: C, 66.49; H, 5.68; N, 11.63. Found: C, 66.57; H, 5.46; N, 11.33.

1-Acetoxyethyl-5-methyl-9-methoxy-6H-pyrido[4,3-b]carbazole, 7: A solution of 51 mg (0.175 mM) of the carbinol **24**, 230 μ l of pyridine and 230 μ l of acetic anhydride in 20 mL of CHCl₃ was stirred at room temperature overnight. The solution was washed thoroughly with H₂O. After drying the solution was concentrated to dryness to leave a red oil which, after chromatography on silica gel (ethyl acetate/hexane, 2:1), gave 26 mg (44%) of the acetate **7**, m.p. 240°C (dec.). IR (KBr) λ 3460-2800, 1735, 1600, 1490, 1415, 1225, 1180, 1035, 815, 775 cm⁻¹; NMR (DMSO- d_6) δ 8.42 (1H, s, H₁₁), 8.37 (1H, d, J=6.0, H₃), 8.19 (1H, s), 7.71 (1H, d, J=6.0, H₄), 7.54 (1H, s, NH), 7.30 (1H, d, J=8.8, H₇), 7.11 (1H, d, J=8.8, H₈), 5.80 (2H, s, CH₂), 3.91 (3H, s, OCH₃), 2.58 (3H, s, OCH₃), 2.16 (3H, s, OCOCH₃), 1.82 (s, H₂O); MS *m/e* 335 (M+1).

Anal. Calcd. for C₂₀H₁₈N₂O₃·0.5 H₂O: C, 69.07; H, 5.53; N, 8.17. Found: C, 70.22; H, 5.57; N, 8.17.

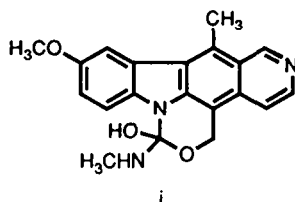
1-Acetoxyethyl-5-methyl-6H-pyrido[4,3-b]carbazole, 25: Acetylation of **23** gave **25** in 46% yield, m.p. 228-230°C. IR (KBr) λ 3400-2800, 1730, 1600, 1470, 1415, 1375, 1235, 1030, 915, 870, 830, 745, 730 cm⁻¹; NMR (CDCl₃) δ 8.59 (1H, s), 8.41 (1H, d, J=6.2), 8.17 (1H, d, J=2.8), 8.13 (1H, d, J=4.4), 7.80-7.18 (2H, m), 5.82 (2H, s, CH₂), 2.71 (3H, s, CH₃), 2.14 (3H, s, COCH₃), 1.6 (s, H₂O); MS *m/e* 305 (M+1).

Anal. Calcd. for C₁₉H₁₆N₂O₂·0.4 H₂O: C, 73.26; H, 5.39; N, 8.99. Found: C, 73.13, H, 5.35; N, 8.82.

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The presence of the 9-methoxy group in **4** renders the indole nitrogen more nucleophilic thereby assisting in the attack on the carbamate carbonyl. An analogous thermolabile isomer was not observed in the synthesis of **1** presumably due to the absence of a 9-methoxy function.

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